

Ammonia As Hydrogen Carrier to Unlock the Full Potential of Green Renewables

Dr Camel Makhloufi – ENGIE Lab CRIGEN
Key Expert – P2X R&D program Leader

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camel.makhloufi@engie.com

ENGIE: Leader in Energy Transition



€189 m
spent on
research and
development (R&D)
in 2019



+ 3 GW
installed
renewables capacity
in 2019



€10.7 bn
revenues in
portfolio for
customer solutions
in 2019



+4,500 km
natural gas
transmission
network
in 2019



52.7GW
natural gas installed
capacity at the close
of 2019

no.1

independent power
producer (IPP) in the world

no.1

Natural gas distribution
network in Europe

no.2

Global provider of
technical installation services

€60.1 bn
revenues

96.8GW
Installed
capacity
of which 26.9 GW
renewable (28%)


7.4GW


16.3GW


2.6GW


39,400 km
natural gas
transmission


171,100
employees

ENGIE: Leader in Energy Transition

Industrial-Scale Projects

Project	Sector
HyNetherlands	Chemical feedstock, industrial fuel, and transport
HyGreen Provence	Mobility and industry
Zev	Mobility
Yuri	Green ammonia
HyEx	Ammonia nitrate
Multiplhy	Bio refinery
Rhyno	Mining
H2@Rail	Rail
Liquid H2	Maritime and more



Demonstrator Projects



Ammonia, a promising and dispatchable energy storage

CO₂ emission and RE production

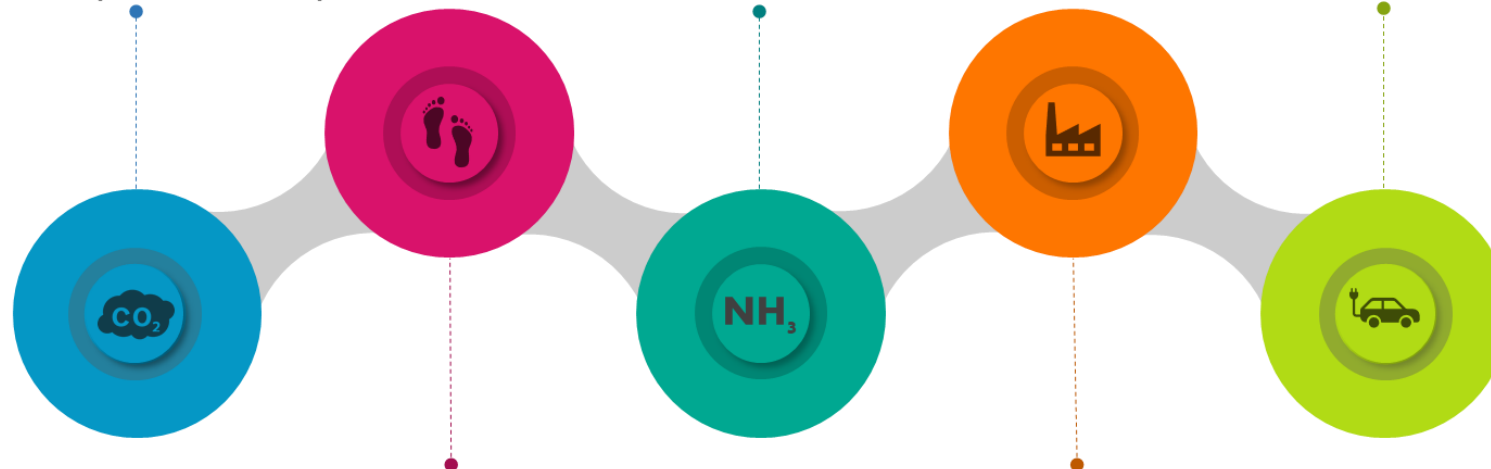
- Power, heat & transport are the main contributors
- Diversification of energy supply through new energy corridors is of utmost importance for Europe

Ammonia as promising solution

- Carbon free e-fuel
- Robust transport & distribution infrastructure
- Existing regulation and safety standards

New technologies for e-NH₃ energy discharge

- Intensified cracker with low CRM impact for decentralized and onboard hydrogen/power generation



E-fuels can unlock hard-to-abate sectors

- EU policies promoting biofuels and e-fuels are implemented
- E-fuels have the least land and water footprints and can be produced at scale

New Technologies for e-NH₃ synthesis

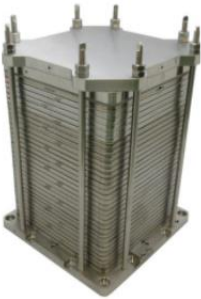
- Flexible NH₃ synthesis loop to cope with intermittent electricity sourcing

Cracking technologies are needed to unleash the full potential of ammonia



Hydrogen refuelling station

- Hydrogen delivered from cracker should have a purity compatible with ISO-14687 with NH_3 max content of 0,1 ppm
- **Cracking system delivering a very pure hydrogen is required at low temperature**
- Fossil fuel Ammonia (and blue ammonia?) comes with various impurities that may be hurt the PEM fuel cell



Fuel cell for decentralized power generation

- Possibly using PEM, SOEC and Alkaline fuel cell with various resistance toward NH_3
- **SOFC can possibly profit from high temperature cracking and offer higher efficiency.**
- Impurities in initial fuel should be a concern



On-board generation for Internal combustion engine

- On-board hydrogen generation may avoid the use of fossil fuel based blend.
- Impact of nitrogen on combustion performance of $\text{NH}_3/\text{H}_2/\text{N}_2$ mixture
- Cracker need to comply with variation of load. Autothermal reformer proposed in some studies

Contaminant	Content	ISO-14687
Carbon Dioxide (CO_2) ¹	≤10 ppm	2 ppm
Chloride (Cl) ¹	≤0.3 ppm	0,05 ppm
Total Sulfur (as SO_4) ¹	≤0.8 ppm	0,004 ppm
Silicon (SiO_2) ²	≤10 ppm	
Heavy metals	ppb to ppm level	

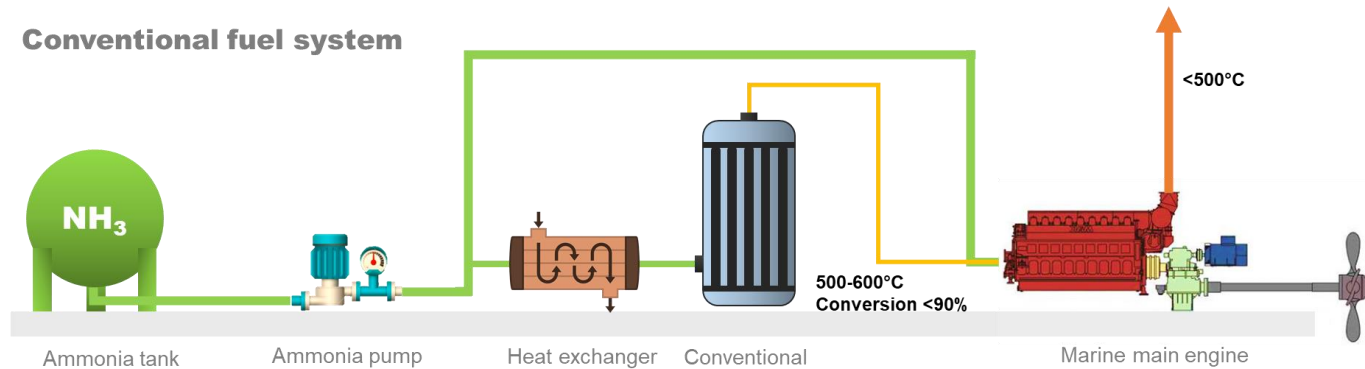
¹ Microelectronic grade

² Technical grade liquor ammonia

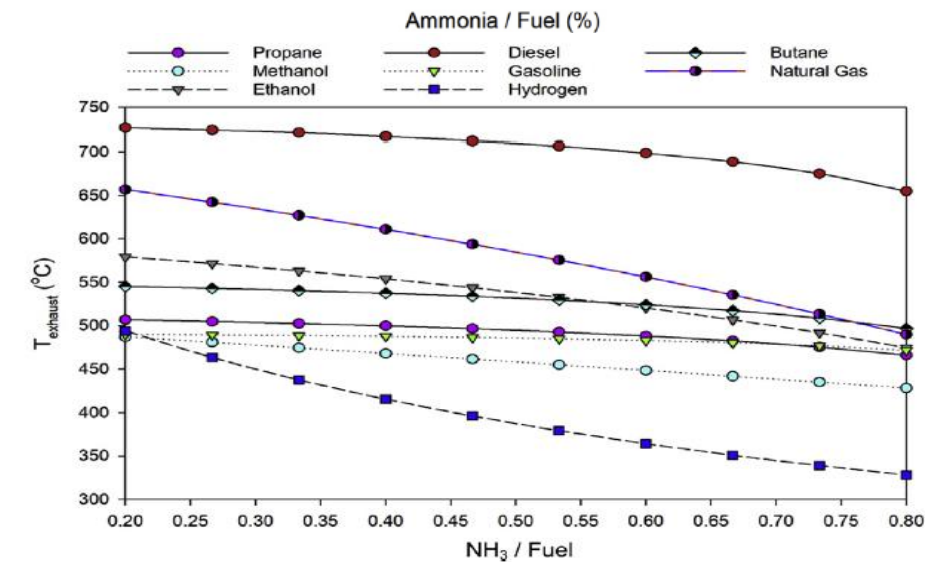
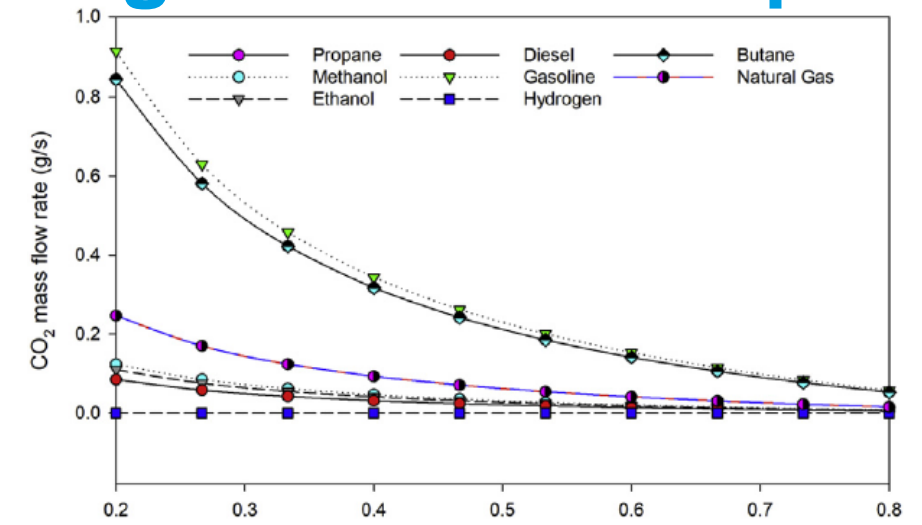
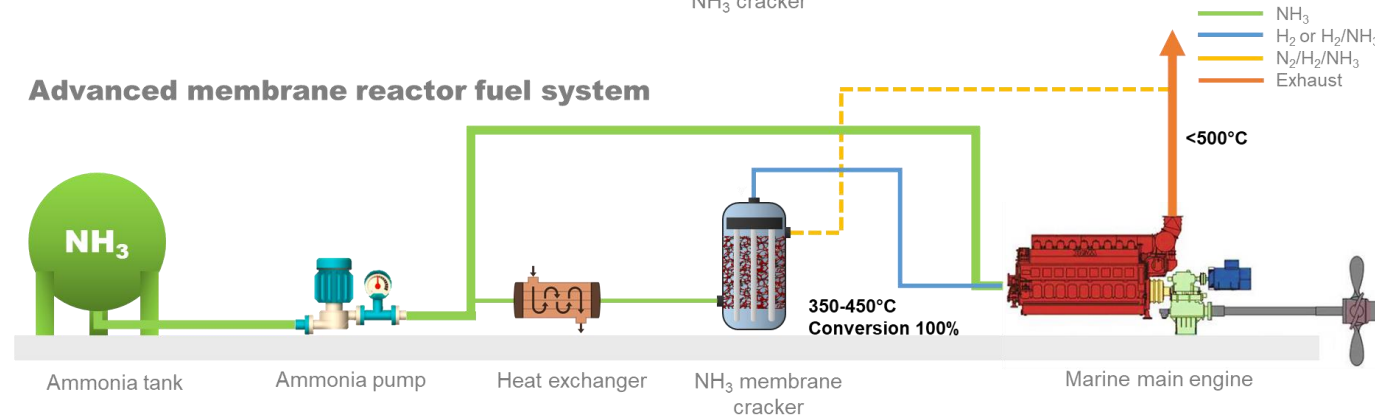
An additional upside of green ammonia: a cleaner fuel for further energy discharge in fuel cells.

On-board ammonia cracking for large ammonia ship

Conventional fuel system



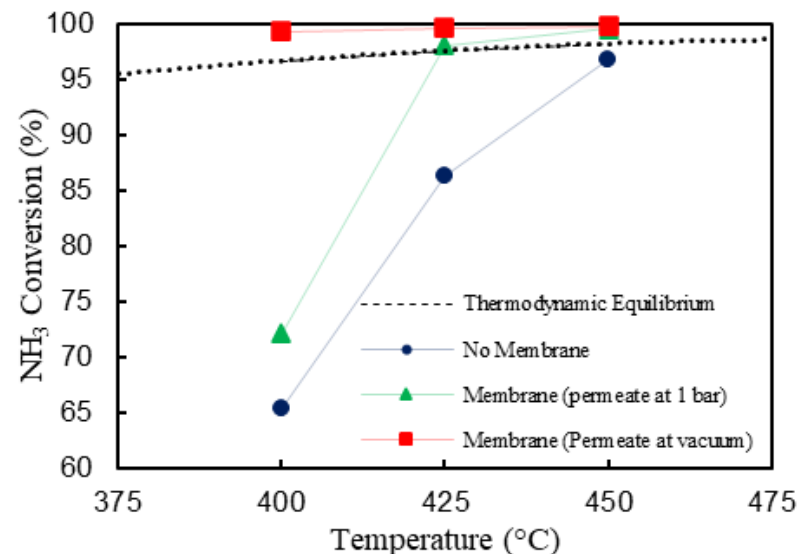
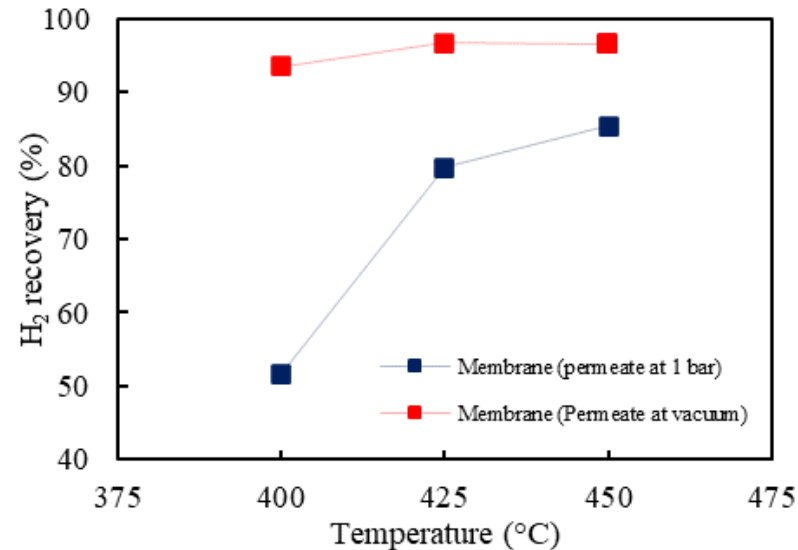
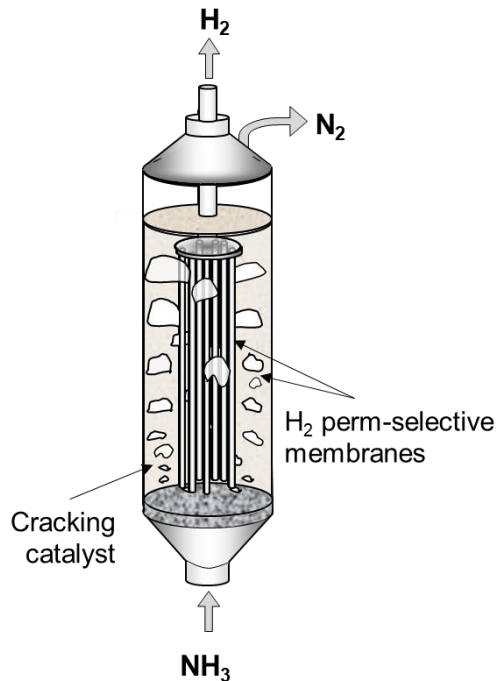
Advanced membrane reactor fuel system



Performance assesment of hydrogen and ammonia combustion with various fuels for power generators; Arda Yapicioglu*, Ibrahim Dincer; 2018

Decentralized Ammonia Reformer : Membrane reactors

Membrane reactor for NH_3 cracking



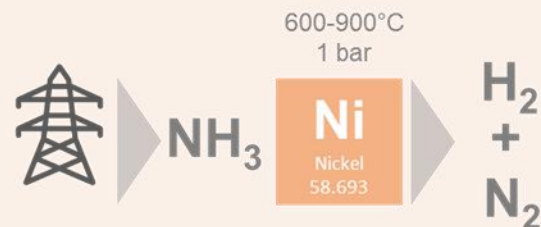
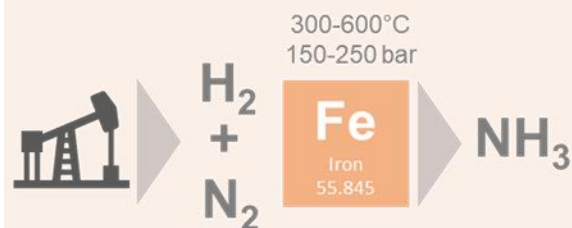
- ❑ In a conventional reactor, the NH_3 conversion stays far from the equilibrium at low temperatures.
- ❑ In a membrane reactor, the conversion is clearly increased and it can reach equilibrium conditions.
- ❑ When vacuum is used in the permeate, virtual full conversion of NH_3 is obtained, even at 400 °C, and in all the cases beyond equilibrium.

H_2 PRODUCTION VIA AMMONIA DECOMPOSITION IN A CATALYTIC MEMBRANE REACTOR; V. Cechetto, L. Di Felice¹, J. Medrano^{1,2}, C. Makhoulfi^{1,3}, J. Zuniga^{1,4}, F. Gallucci¹

Does decentralization imply a high consumption of valuable critical raw materials?

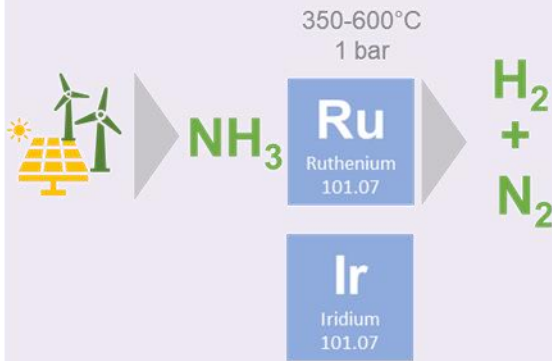
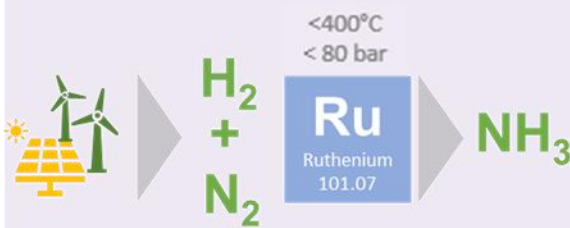
Conventional catalysts

- ☒ Fossil fuel feedstock
- ☒ Harsh reaction conditions
- ☒ Low process flexibility
- ☒ Low catalyst activity
- ☒ Abundant and cheap materials



Alternative catalysts

- ☒ Renewable feedstock
- ☒ Mild reaction conditions
- ☒ Higher process flexibility
- ☒ Higher catalyst activity
- ☒ Rare and expensive materials



Tomorrow's catalysts

- ☒ Renewable feedstock
- ☒ Mild reaction conditions
- ☒ High process flexibility
- ☒ High catalyst activity
- ☒ Non-transition metals

