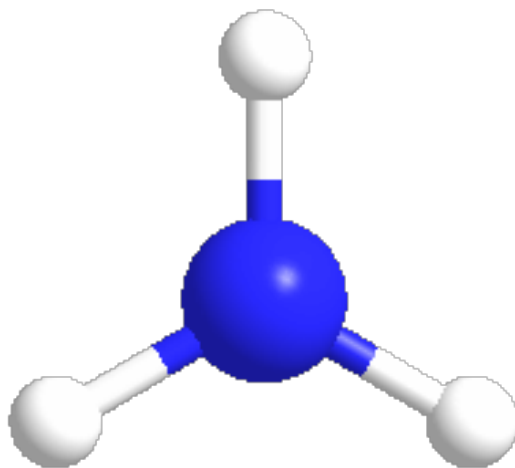


Safety of Ammonia As Hydrogen and Energy Carriers



Wednesday, November 13, 2019 10:20-10:38

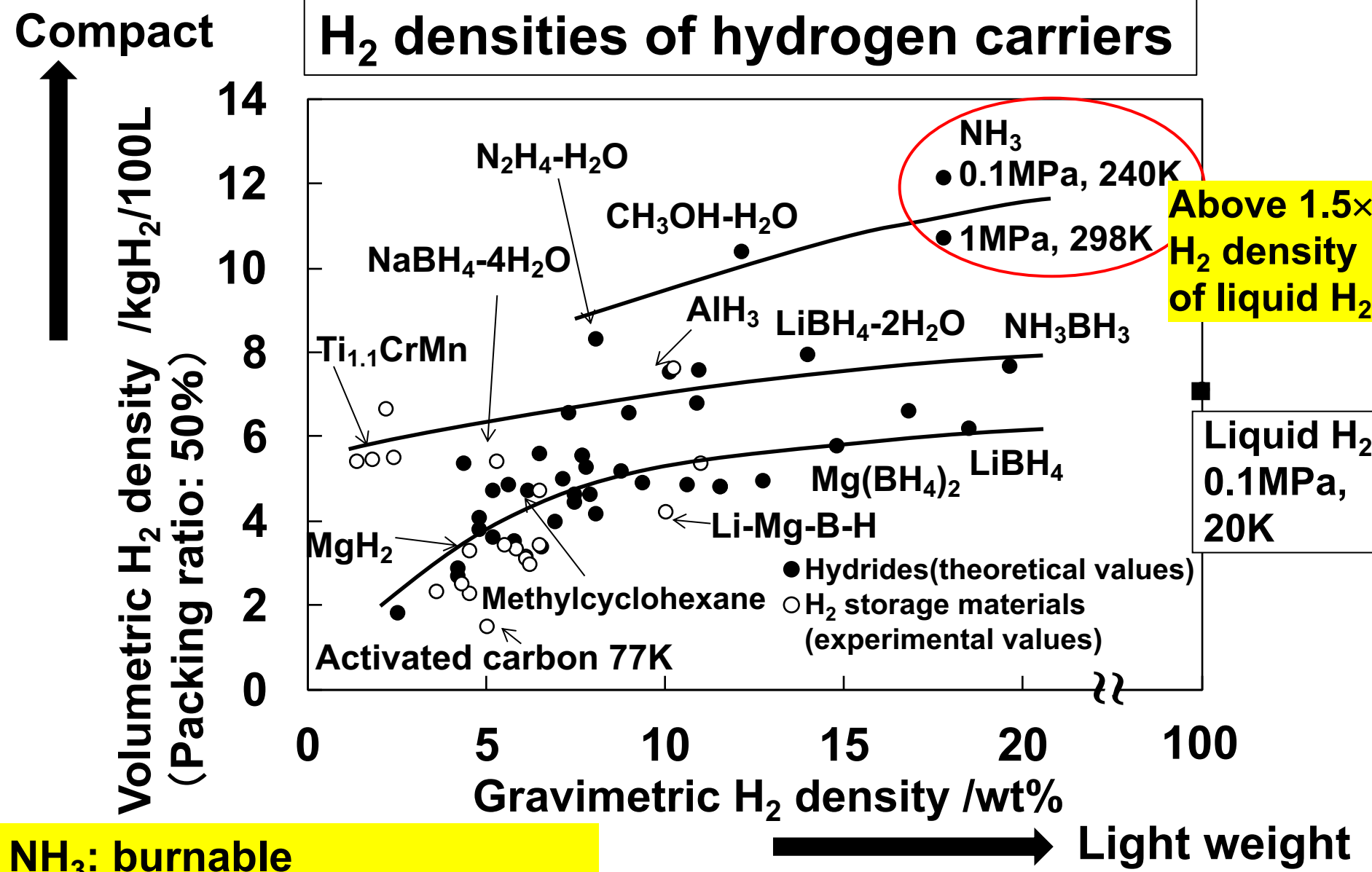
Hyatt Regency Orlando - Regency Ballroom P

**Ammonia Energy Conference 2019 at 19 AIChE Annual
Meeting in Orlando, FL, November 12-14, 2019**

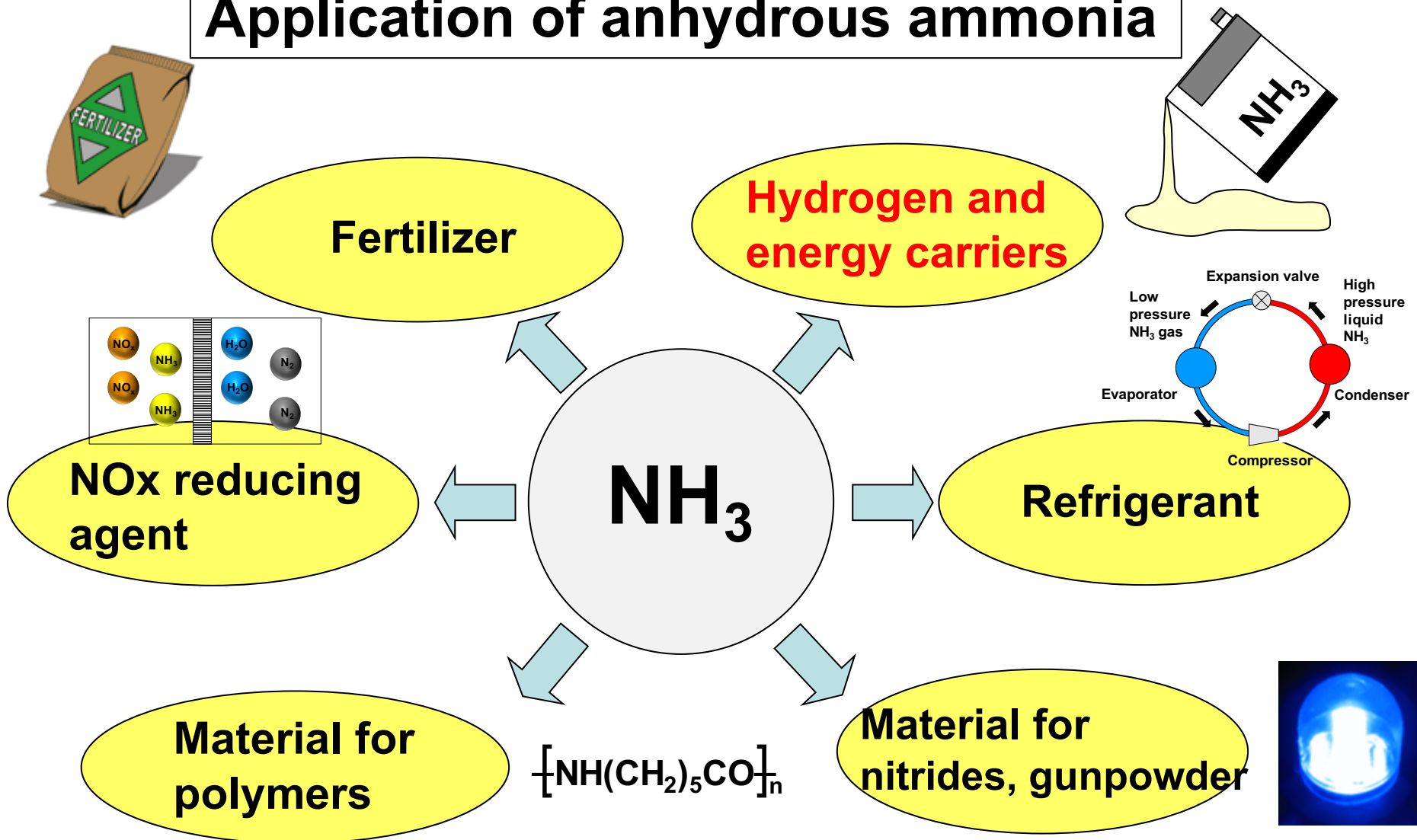
Yoshitsugu Kojima

**Natural Science Center for Basic Research and
Development, Hiroshima University, Japan**

1. Introduction



Application of anhydrous ammonia



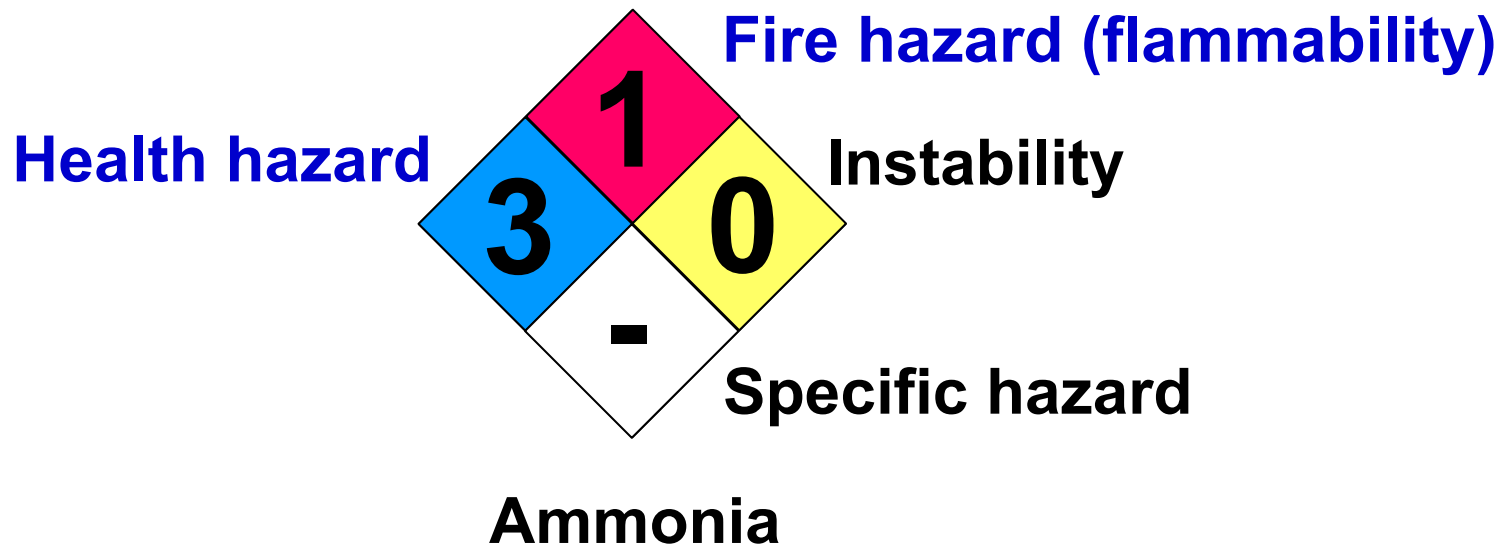
Ammonia: deleterious substance

Purpose: To figure out regulations for safety of NH_3 , survey NH_3 accident and propose new concept of ammonia removal system

2. Regulations for safety of NH₃

(1)USA: NFPA(National Fire Protection Association) 704

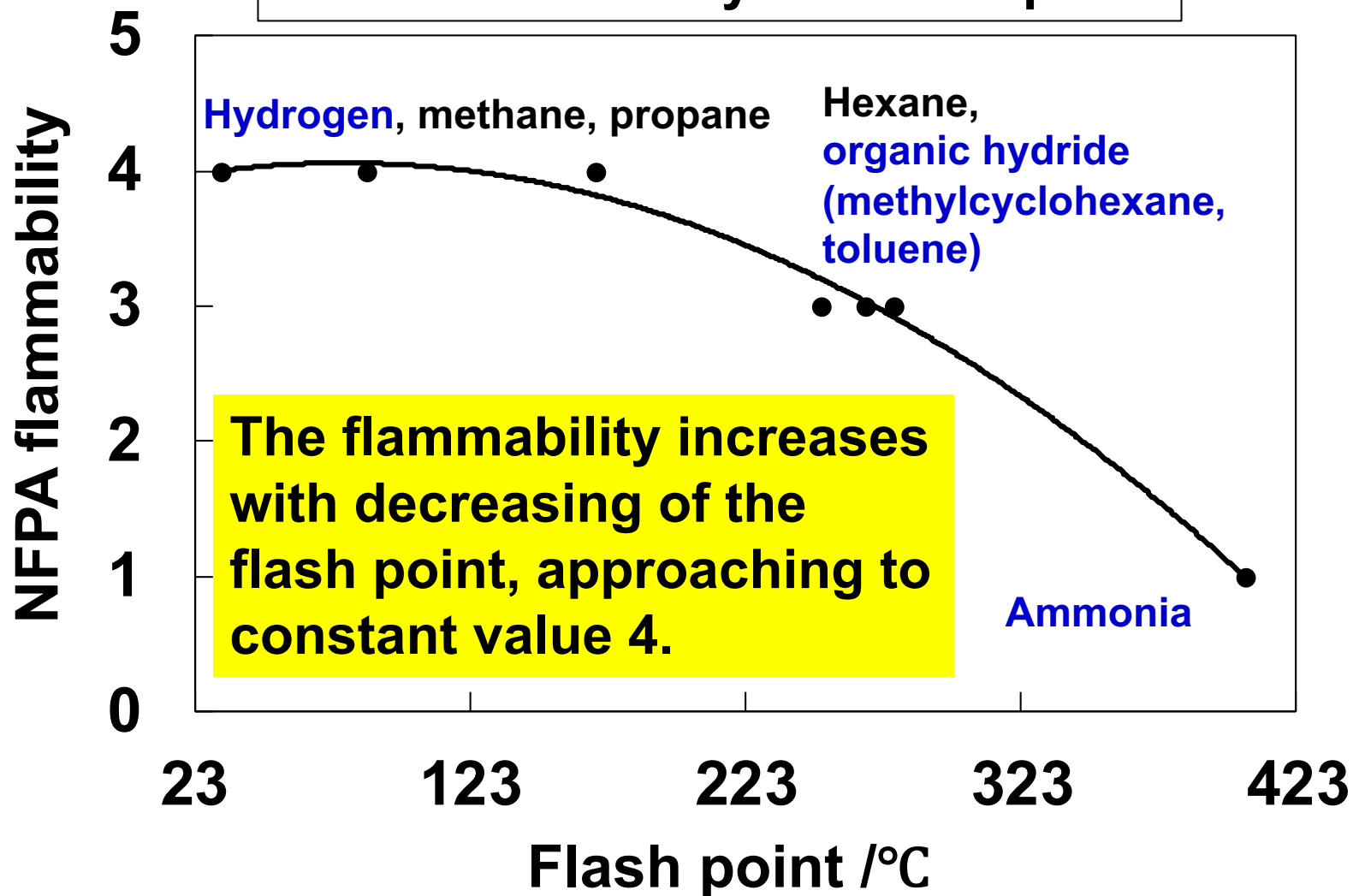
Fire diamond of ammonia



(2)Japan: GHS (The Globally Harmonized System of Classification and Labelling of Chemicals)

(3)EU: CLP regulation (Regulation on Classification, Labelling and Packaging of substances mixtures)

NFPA flammability and flash point

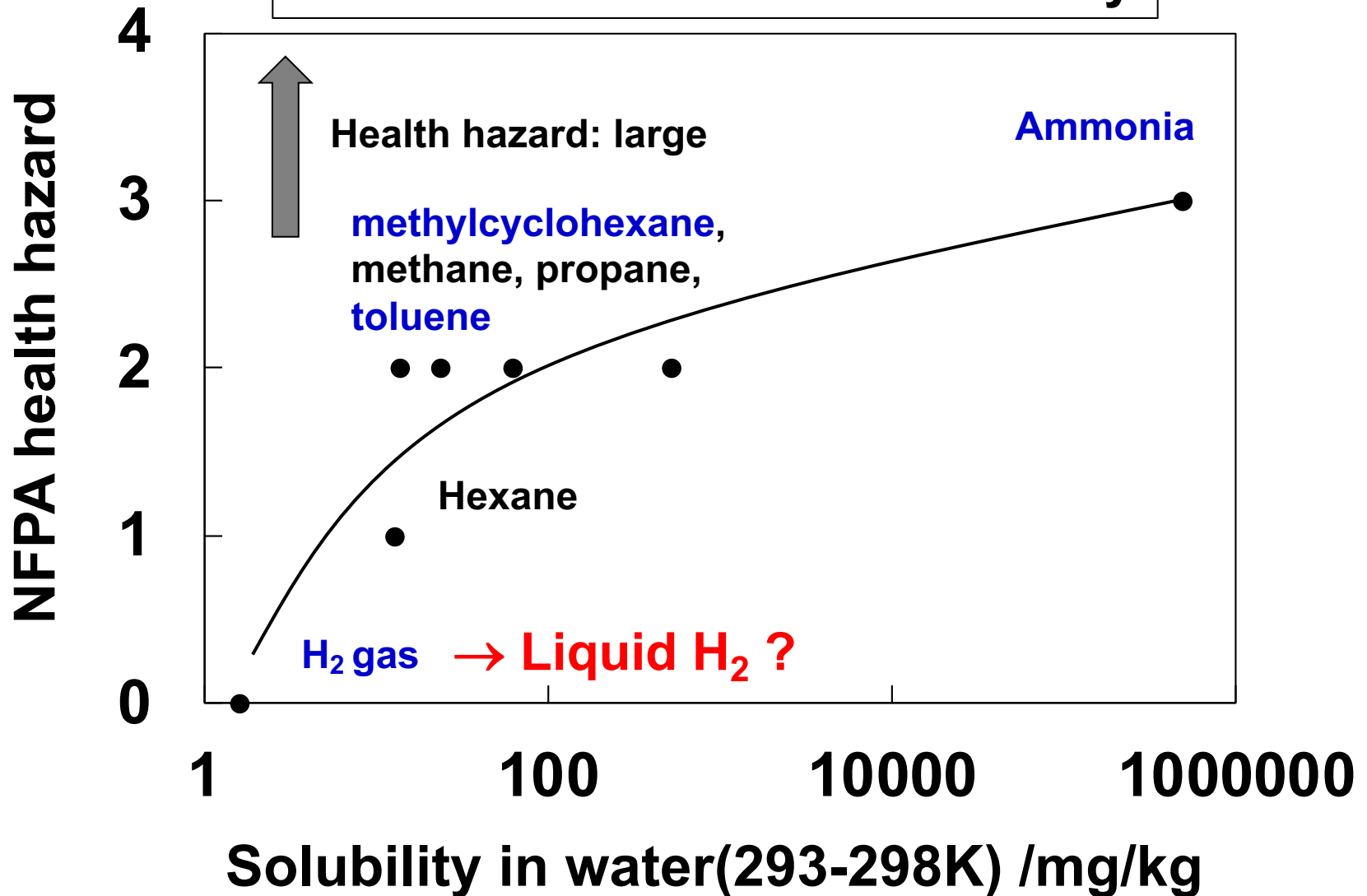


Fire hazard: **Ammonia** < Organic hydrides < Hydrogen

Japanese GHS

Fire hazard: Organic hydrides < Hydrogen, **Ammonia**

NFPA health hazard and solubility



The health hazard increases with the solubility in water.

Health hazard: Hydrogen gas < Organic hydrides < Ammonia

Fire diamond of hydrogen and energy carriers

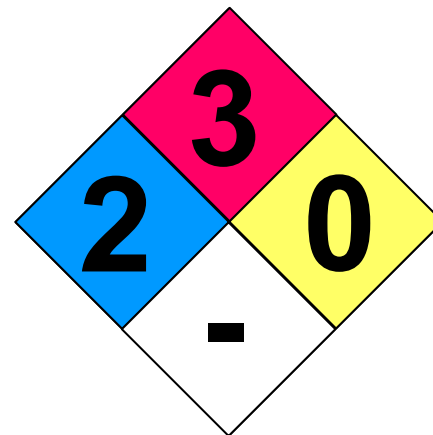
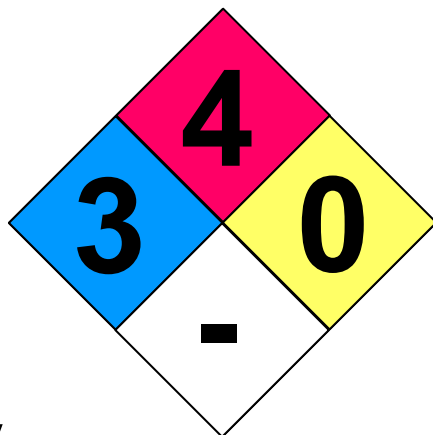
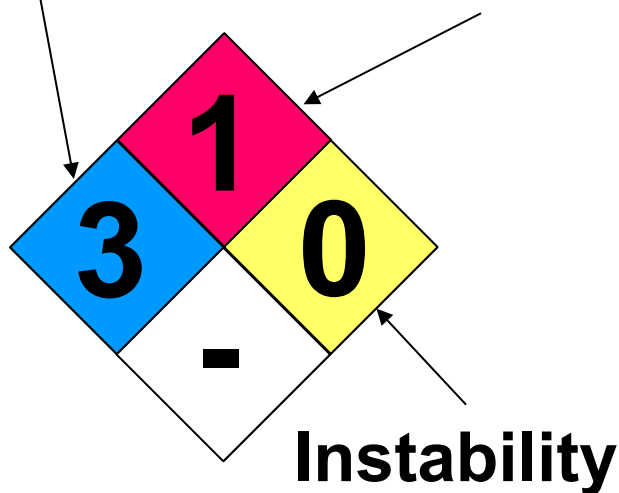
Ammonia (NH₃)

Liquid hydrogen

Organic hydrides
(Methylcyclohexane,
Toluene)

Health hazard

Fire hazard
(flammability)

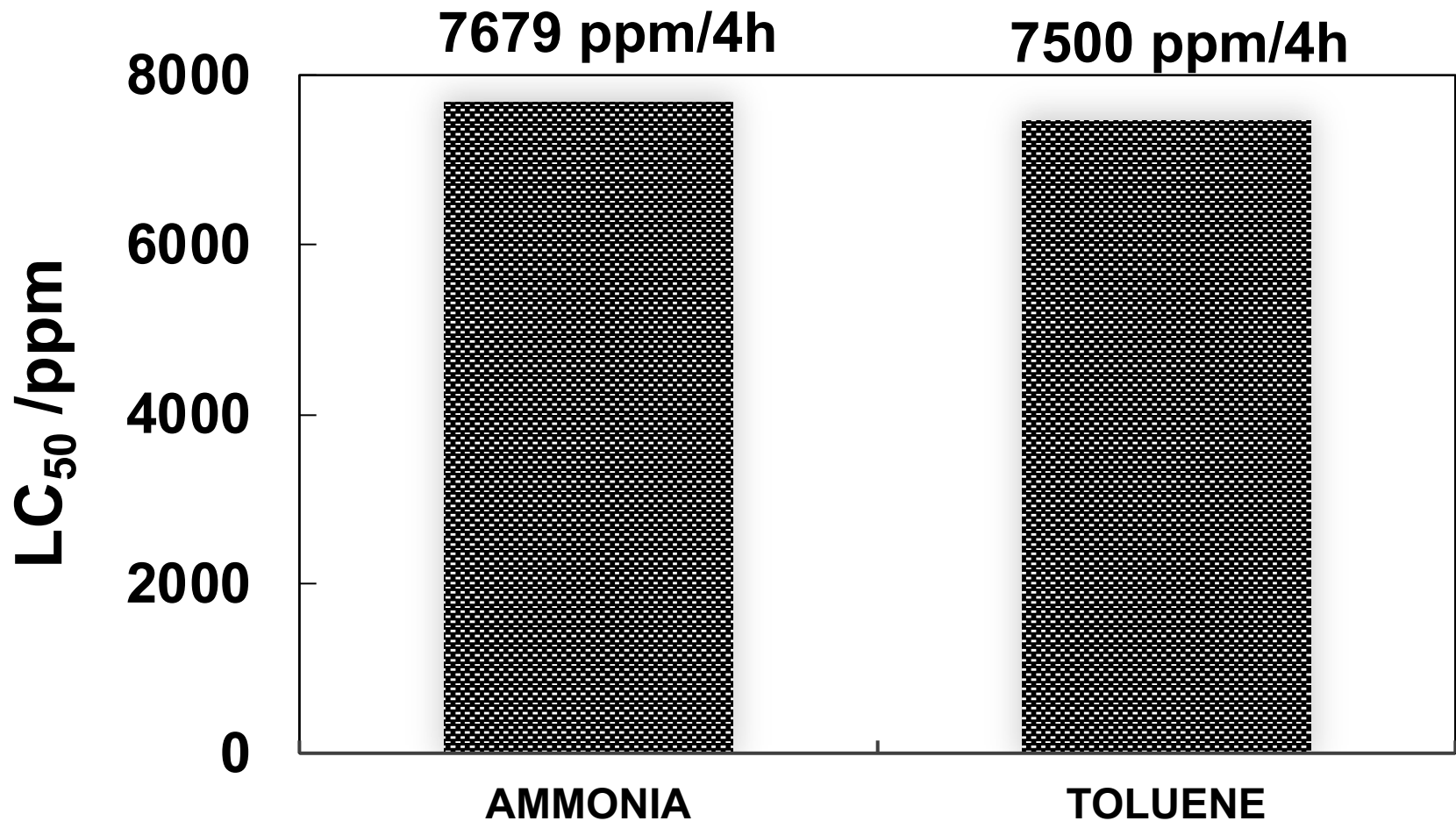


Ammonia and liquid H₂ have same health hazard.

Safety:

Liquid hydrogen(7) < Organic hydrides(5) < Ammonia(4)

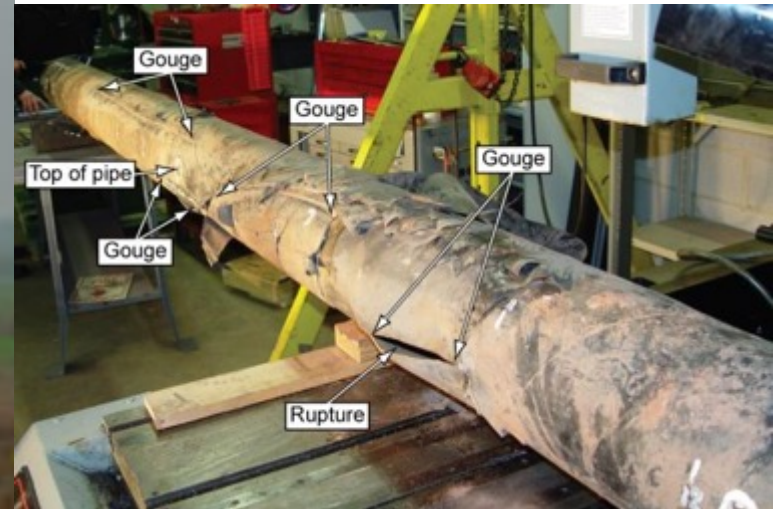
LC₅₀ (50% lethal concentration, 4h rat inhalation) of ammonia and toluene



Ammonia and toluene have same health hazard.

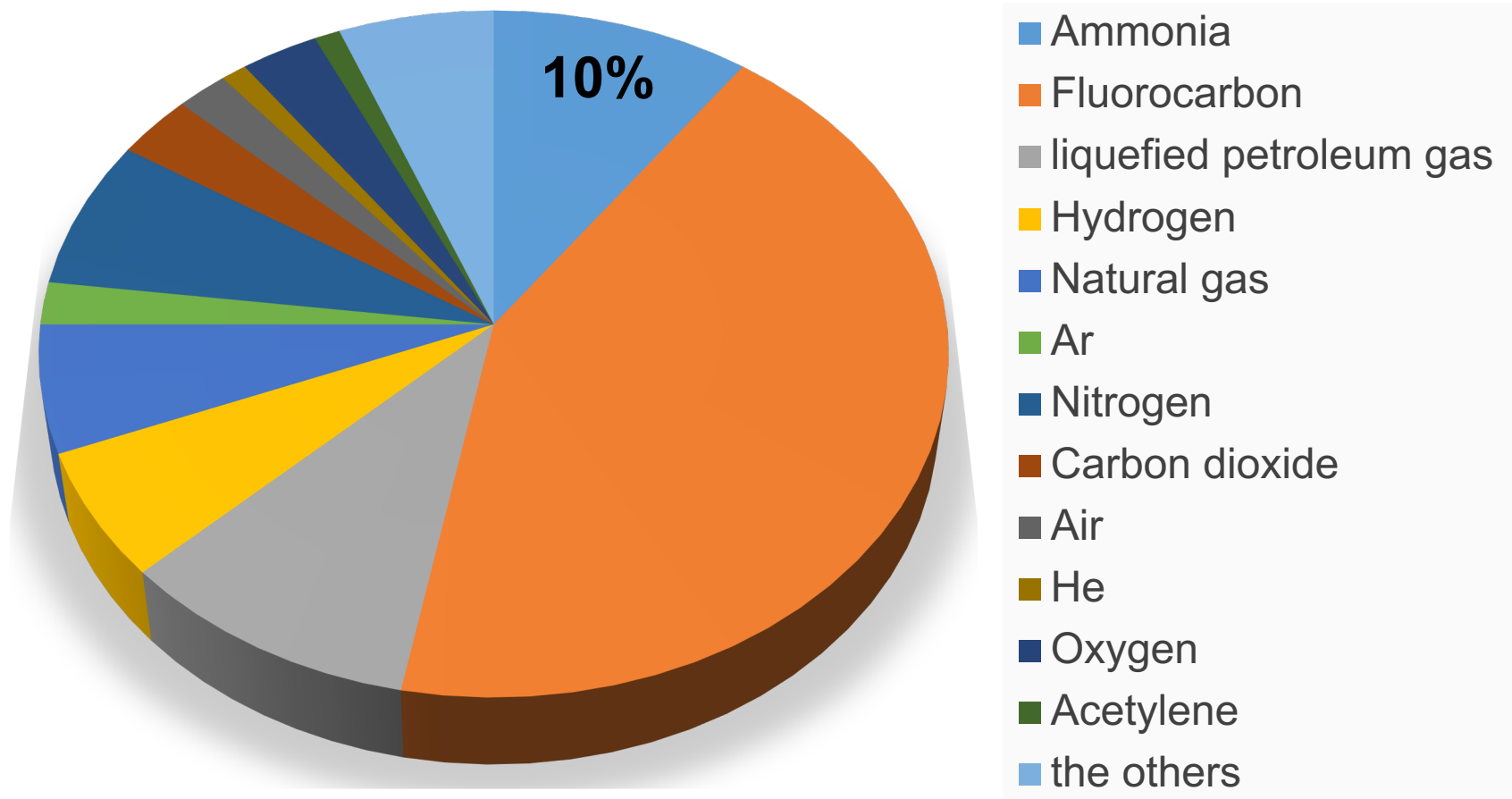
3. Survey of NH₃ accident

Leakage from the ammonia pipeline, Kansas, USA



October 27, 2004

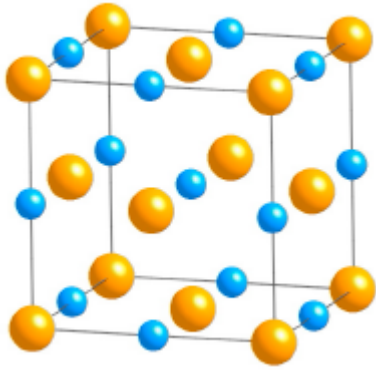
The ratio of the number of ammonia accidents to the number of high pressure gas related accidents in Japan (2014)



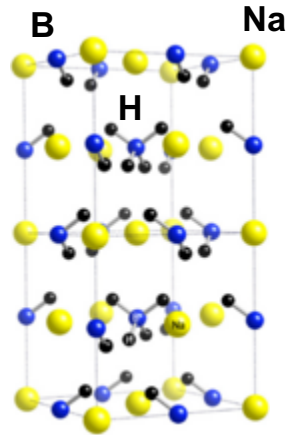
Caused by the leakage from valve, pipe and flange

4. New concept of ammonia removal system

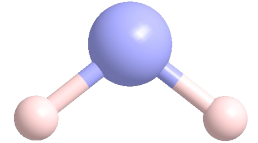
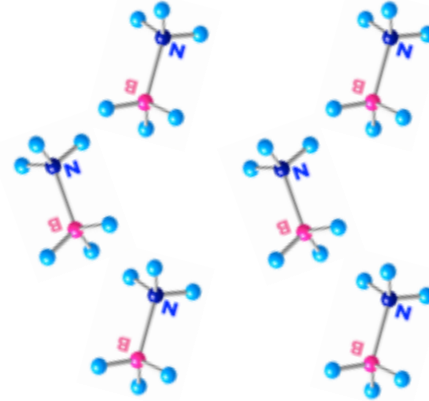
Ammonia storage (absorption, adsorption) materials



Metal halides

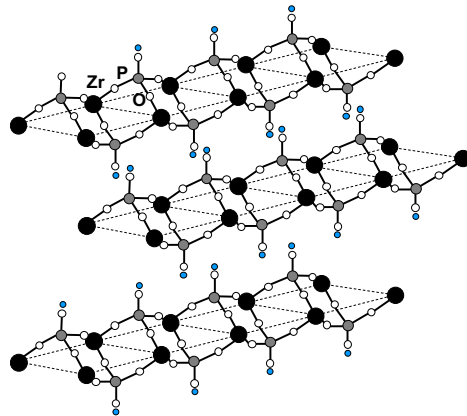
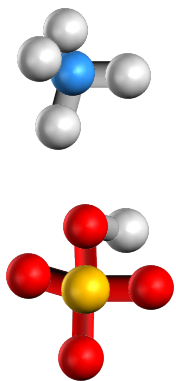


Complex hydrides

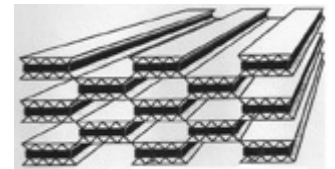
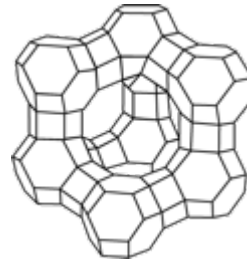


Water

Ammonia remover



Proton-based materials

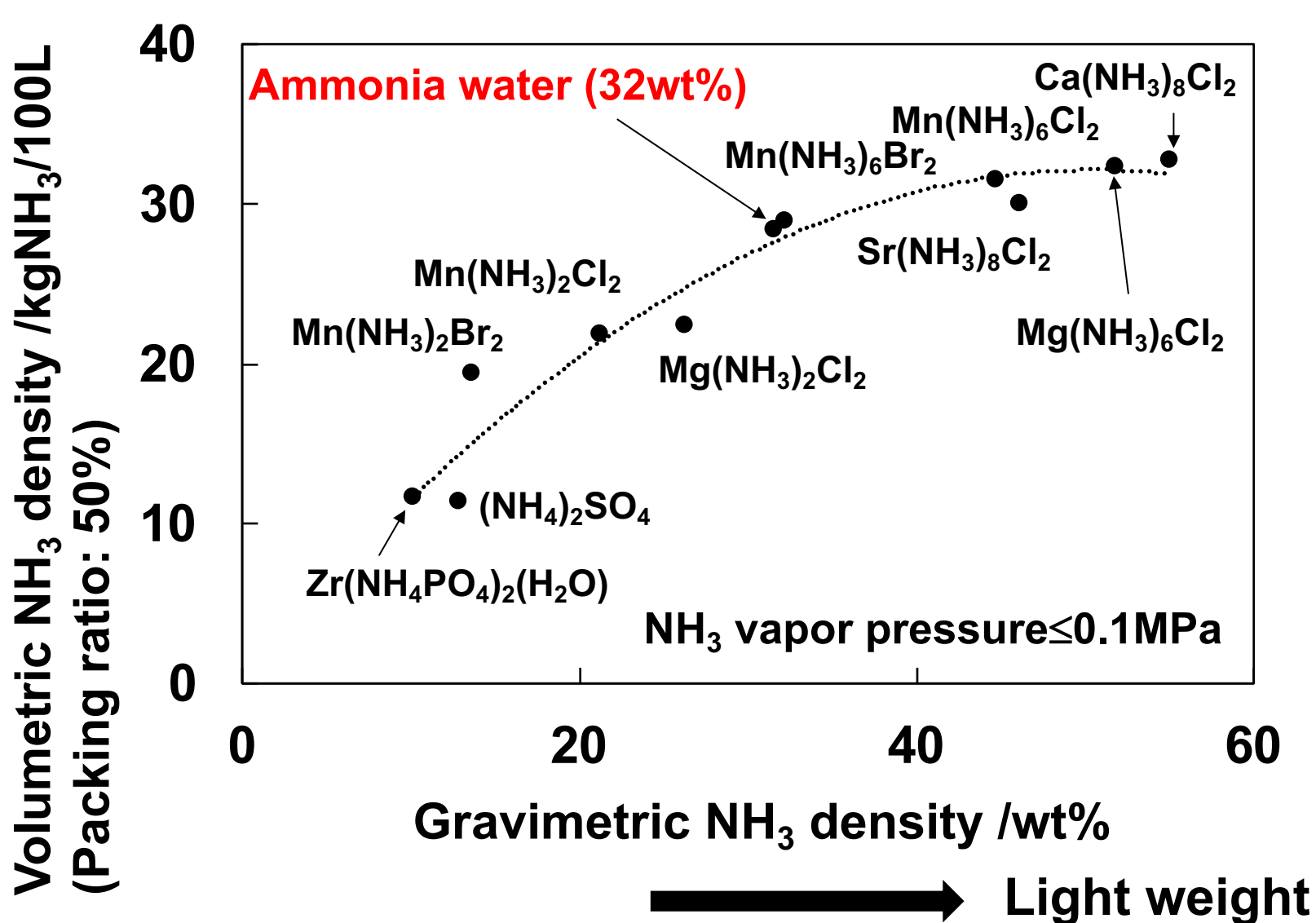


Porous materials

Evaluation and characterization ammonia storage materials

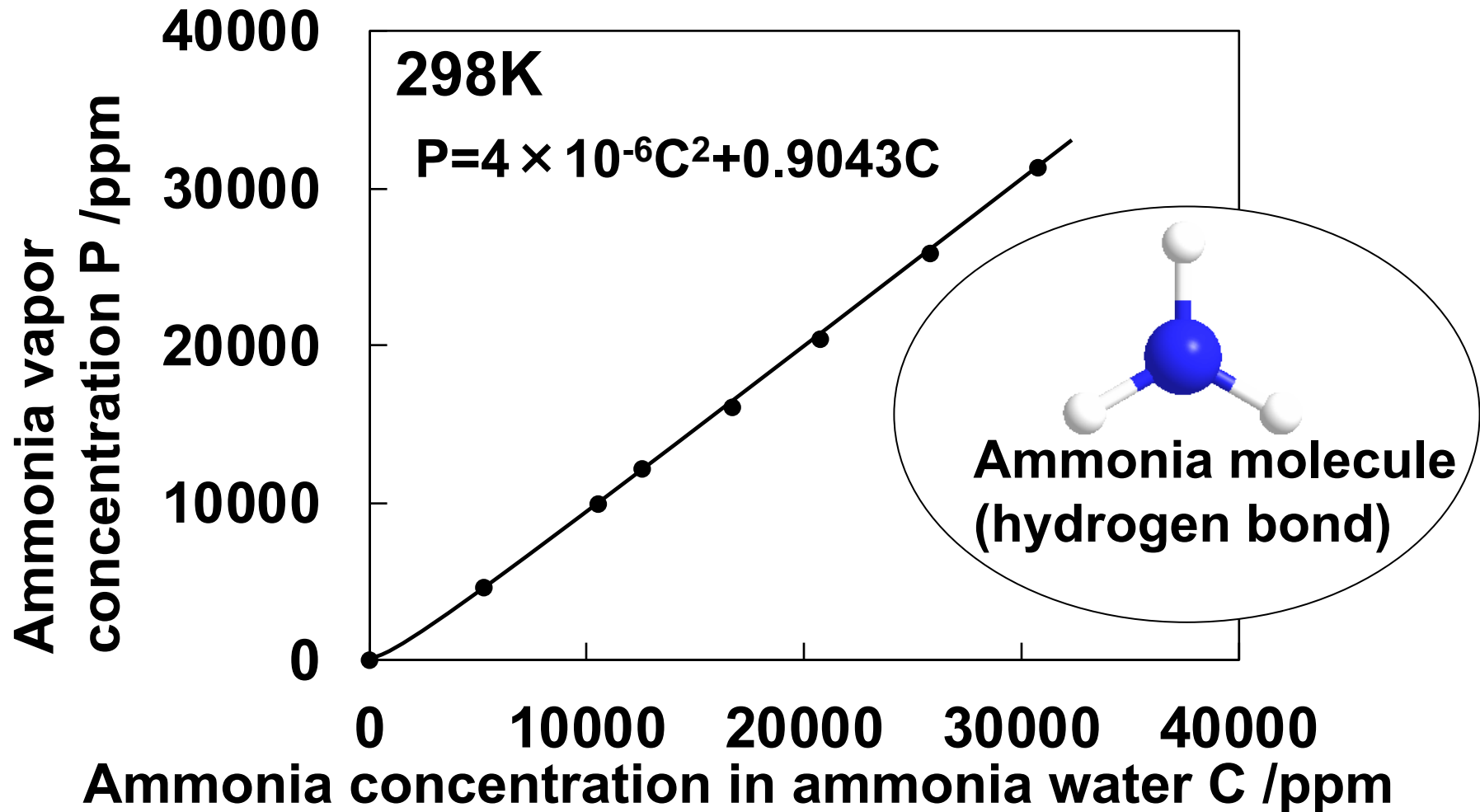
Compact

NH₃ densities of NH₃ storage materials



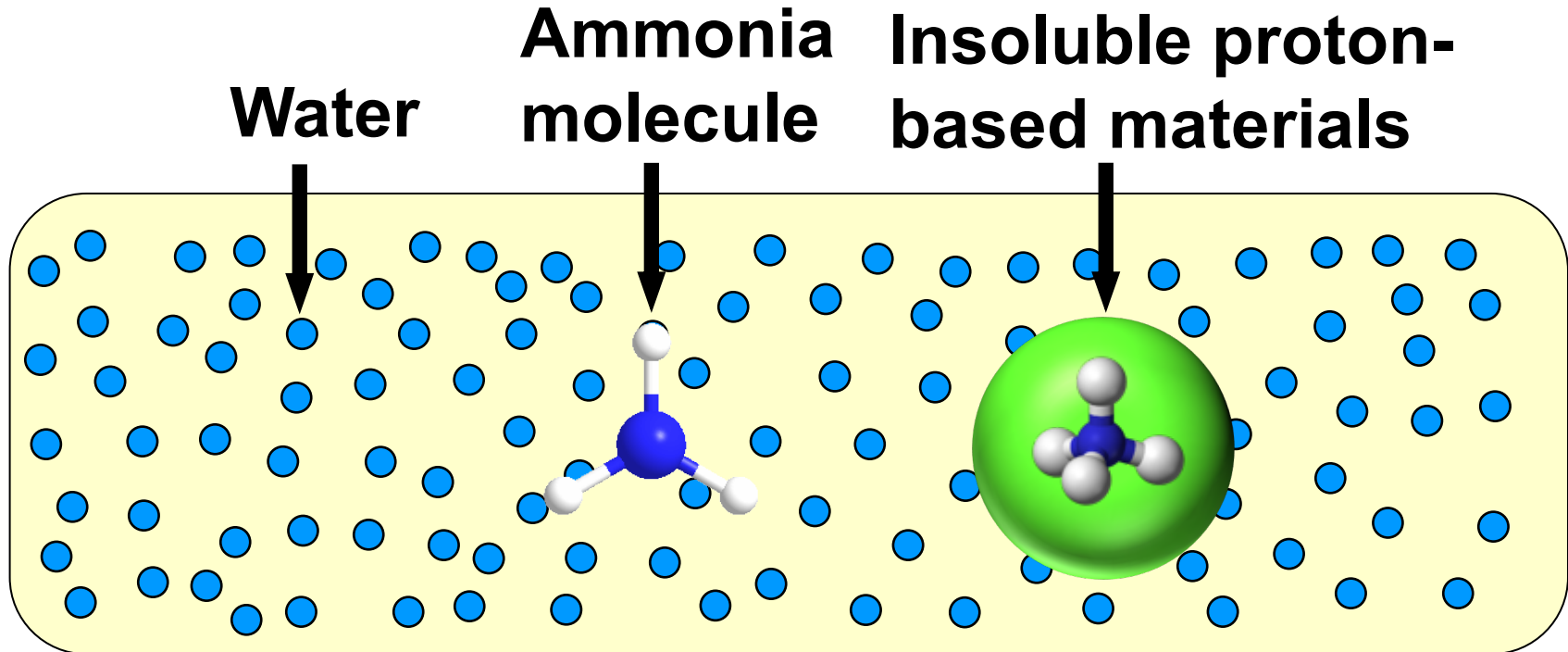
NH₃: large diffusion coefficient in water

Ammonia vapor concentration and ammonia concentration in ammonia water



NH₃ concentration: 2500ppm → NH₃ vapor concentration: 2300ppm
(To decrease NH₃ vapor concentration → Insoluble proton-based materials)

Conceptive picture of ammonia removal system



**Insoluble ammonium ion-based materials
(coordinate bond) to remove nitrogen**

The NH_3 removal system combined water and Insoluble proton-based materials will be candidate to reduce NH_3 concentration in the atmosphere by suppression of water pollution.

5. Summary

- (1) NH_3 is safer than liquid hydrogen and organic hydrides by addition of health hazard and fire hazard based on fire diamond.**
- (2) Water absorbs large amount of NH_3 , although the vapor concentration is high.**
- (3) The NH_3 removal systems combined water and insoluble proton-based materials will be candidate to reduce leaked NH_3 in the atmosphere.**

Acknowledgements

This work was supported by Council for Science, Technology and Innovation(CSTI), Cross-ministerial Strategic Innovation Promotion Program (SIP), “energy carrier”(funding agency : JST).

Thank you for your attention.